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oxidation $CO \rightarrow CO_2 + 2e^-$ reduction $Fe_2O_3 + 2e^- \rightarrow Fe$ 24 3 $AgNO_3 + Ni \rightarrow Ni(NO_3)_2 + Ag$ oxidation reduction 25 $Cu_2O + Pb \rightarrow PbO + Cu$ oxidation reduction 26 O_2

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 $5CO + i_2O_5 \rightarrow 5CO_2 + I_2$ $5CO + I_2O_5 \rightarrow 5CO_2 + I_2$

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