

## CONSTANTES, UNITÉS ET SYMBOLES USUELS

Quelques constantes fondamentales (IUPAC 2015)			
Grandeur physique	Symbole	Valeur	Unités
constante d'Avogadro	$N_A$	$6,02214179 \cdot 10^{23}$	$\text{mol}^{-1}$
vitesse de la lumière dans le vide	$c, c_0$	$2,99792458 \cdot 10^8$	$\text{m} \cdot \text{s}^{-1}$
accélération de la pesanteur	$g$	9,80665	$\text{m} \cdot \text{s}^{-2}$
charge élémentaire	$e$	$1,602176487 \cdot 10^{-19}$	C
volume molaire du gaz parfait à T.P.N.	$V_m$	$2,241410 \cdot 10^{-2}$	$\text{m}^3 \cdot \text{mol}^{-1}$
et sous 1 bar à 273,15 K	$V_m^0$	$2,271108 \cdot 10^{-2}$	$\text{m}^3 \cdot \text{mol}^{-1}$
constante des gaz	$R$	8,314472	$\text{J} \cdot \text{K}^{-1} \cdot \text{mol}^{-1}$
	$R$	0,08205783	$\text{L} \cdot \text{atm} \cdot \text{K}^{-1} \cdot \text{mol}^{-1}$
constante de Boltzmann	$k, k_B$	$1,3806504 \cdot 10^{-23}$	$\text{J} \cdot \text{K}^{-1}$
constante de Planck	$h$	$6,62606896 \cdot 10^{-34}$	J·s
constante de Faraday	$F$	$9,64853399 \cdot 10^4$	$\text{C} \cdot \text{mol}^{-1}$
constante de Rydberg	$R_\infty$	$1,0973731534 \cdot 10^7$	$\text{m}^{-1}$
		$2,1798741 \cdot 10^{-18}$	J
unité de masse atomique	$m_u, u$	$1,6605402 \cdot 10^{-27}$	kg
masse de l'électron au repos	$m_e$	$9,1093897 \cdot 10^{-31}$	kg
masse du proton au repos	$m_p$	$1,6726231 \cdot 10^{-27}$	kg
masse du neutron au repos	$m_n$	$1,6749286 \cdot 10^{-27}$	kg
rayon de Bohr	$a_0$	$5,29177249 \cdot 10^{-11}$	m
perméabilité du vide	$\mu_0$	$4\pi \cdot 10^{-7}$	$\text{N} \cdot \text{A}^{-2}$
permittivité du vide	$\epsilon_0$	$8,854187817 \cdot 10^{-12}$	$\text{F} \cdot \text{m}^{-1}$

Alphabet grec					
Nom	minuscule	majuscule	Nom	minuscule	majuscule
Alpha	$\alpha$	Α	nu	$\nu$	Ν
bêta	$\beta$	Β	xi	$\xi$	Ξ
gamma	$\gamma$	Γ	omicron	$\omicron$	Ο
delta	$\delta$	Δ	pi	$\pi$	Π
epsilon	$\epsilon$	Ε	rho	$\rho$	Ρ
zêta	$\zeta$	Ζ	sigma	$\sigma$	Σ
êta	$\eta$	Η	tau	$\tau$	Τ
thêta	$\theta$	Θ	upsilon	$\upsilon$	Υ
iota	$\iota$	Ι	phi	$\phi$	Φ
kappa	$\kappa$	Κ	chi	$\chi$	Χ
lambda	$\lambda$	Λ	psi	$\psi$	Ψ
mu	$\mu$	Μ	oméga	$\omega$	Ω

# Tables De Constantes

**Robert R. Redfield**



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## **Table of Contents Tables De Constantes**

1. Understanding the eBook Tables De Constantes
  - The Rise of Digital Reading Tables De Constantes
  - Advantages of eBooks Over Traditional Books
2. Identifying Tables De Constantes
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Tables De Constantes
  - User-Friendly Interface
4. Exploring eBook Recommendations from Tables De Constantes
  - Personalized Recommendations
  - Tables De Constantes User Reviews and Ratings
  - Tables De Constantes and Bestseller Lists
5. Accessing Tables De Constantes Free and Paid eBooks
  - Tables De Constantes Public Domain eBooks
  - Tables De Constantes eBook Subscription Services

- Tables De Constantes Budget-Friendly Options
- 6. Navigating Tables De Constantes eBook Formats
  - ePub, PDF, MOBI, and More
  - Tables De Constantes Compatibility with Devices
  - Tables De Constantes Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Tables De Constantes
  - Highlighting and Note-Taking Tables De Constantes
  - Interactive Elements Tables De Constantes
- 8. Staying Engaged with Tables De Constantes
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Tables De Constantes
- 9. Balancing eBooks and Physical Books Tables De Constantes
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Tables De Constantes
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Tables De Constantes
  - Setting Reading Goals Tables De Constantes
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Tables De Constantes
  - Fact-Checking eBook Content of Tables De Constantes
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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